

2010 HIGHER SCHOOL CERTIFICATE EXAMINATION

# Chemistry

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Centre Number

## Section I (continued)

Part B – 55 marks

Attempt Questions 21–31

Allow about 1 hour and 40 minutes for this part

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Student Number

Answer the questions in the spaces provided. These spaces provide guidance for the expected length of response.

Show all relevant working in questions involving calculations.

### Question 21 (3 marks)

A 0.001 mol L<sup>-1</sup> solution of hydrochloric acid and a 0.056 mol L<sup>-1</sup> solution of ethanoic acid both have a pH of 3.0. 3

Why do both solutions have the same pH?

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