Question 21 (3 marks)	
A $0.001~\text{mol}~\text{L}^{-1}$ solution of hydrochloric acid and a $0.056~\text{mol}~\text{L}^{-1}$ solution of ethanoic acid both have a pH of 3.0 .	3
Why do both solutions have the same pH?	
They have a motor ratio of 1.1. They are both acidic	